

'2001' Nuclear Quadrupole Moments
 $Q/\text{millibarn} (=10^{-31} \text{ m}^2)$

1 (I)	2 (II)	3	4	5	6	7	8	9	10	11	12	13 (III)	14 (IV)	15 (V)	16 (VI)	17 (VII)	18 (VIII)	
1 <i>m</i> H 2 2.86(15)																	2 - He	
3 <i>m</i> Li 6 -0.808	4 <i>a</i> Be 9 52.88(38)											5 <i>a</i> B 10 84.59(24) 11 40.59(10)	6 <i>a</i> C 11 33.27(24)	7 <i>a</i> N 14 20.44(3)	8 <i>a</i> O 17 -25.58(22)	9 <i>m+s</i> F 19* -94.2(9)	10 <i>a</i> Ne 21 101.55(75)	
7 -40.1													13 <i>a+m</i> Al 27 146.6(10)	14 <i>a</i> Si 28 -	15 <i>a</i> P 31 -	16 <i>a</i> S 33 -67.8(13) 35 47.1(9)	17 <i>a</i> Cl 35 -81.65(80) 37 -64.35(64)	18 <i>a</i> Ar 39 -
11 <i>m</i> Na 23 104(1)	12 <i>a</i> Mg 25 199.4(20)																	
19 <i>m</i> K 39 58.5 40 -73 41 71.1	20 <i>a</i> Ca 41 -66.5(18) 43 -40.8(8)	21 <i>m</i> Sc 45 -220(2)	22 <i>a</i> Ti 47 302(10) 49 247(11)	23 <i>a</i> V 50 210(40) 51 -52(10)	24 <i>a</i> Cr 53 -150(50) 55 330(10)	25 <i>a</i> Mn 55 330(10)	26 <i>s</i> Fe 57 160	27 <i>a</i> Co 59 +420(30)	28 <i>a</i> Ni 61 +162(15)	29 <i>μ</i> Cu 63 -220(15) 65 -204(14)	30 <i>a</i> Zn 67 +150(15) 71 107(1)	31 <i>m</i> Ga 69 171(2) 71 107(1)	32 <i>m</i> Ge 73 -196 75 314(6)	33 <i>μ</i> As 75 314(6)	34 <i>a</i> Se 79 -	35 <i>a+m</i> Br 79 313(3) 81 261.5(25)	36 <i>m</i> Kr 83 +259(1) 85 507(3)	
37 <i>m</i> Rb 85 +276(1) 87 +133.5(5)	38 <i>a</i> Sr 87 335(20) 89 +245(4)	39 <i>a</i> Y 90 -125(11)	40 <i>m</i> Zr 91 -176(3) 93 +3793(33)	41 <i>μ</i> Nb 93 -320(20) 95 3710(70)	42 <i>a</i> Mo 95 -22(1) 97 +255(13)	43 <i>a</i> Tc 99 -129(6)	44 <i>a</i> Ru 99 +79(4) 101 +457(23)	45 <i>a</i> Rh 101 -	46 <i>μ</i> Pd 105 +660(11)	47 <i>as</i> Ag 107 980(110)	48 <i>a</i> Cd 111* -850(90)	49 <i>a</i> In 113 +799 115 +810	50 <i>s</i> Sn 119 -128(7) 121 -480(50) 123 -490(50)	51 <i>a</i> Sb 121 -360(40) 125 -310(20)	52 <i>a</i> Te 125 -310(20)	53 <i>a+m</i> I 127 -710(10) 129 -636(9) 131 -616(9)	54 <i>m</i> Xe 129 -393(10) 131 -114(1)	
55 <i>ms</i> Cs 133 -3.43(10) 135 302(21)	56 <i>a</i> Ba 135 +160(3) 137 +245(4)	La- Lu	72 <i>μ</i> Hf 177 +3365(29) 179 +3793(33)	73 <i>π</i> Ta 181 3170(20) 183 3710(70)	74 <i>a</i> W 182 -2130(350)	75 <i>π</i> Re 185 2180(20) 187 2070(20)	76 <i>μ</i> Os 189 +856(28) 191 +816(9) 193 +751(9)	77 <i>μ</i> Ir 191 +816(9) 193 +751(9)	78 <i>a</i> Pt 195 -	79 <i>μ</i> Au 197 547(16)	80 <i>μ+s</i> Hg 199 +674(77) 201 +386(49)	81 <i>a</i> Tl 203 -	82 <i>a</i> Pb 209 -269(165)	83 <i>a</i> Bi 209 -516(15)	84 <i>a</i> Po 209 -	85 <i>a</i> At 209 -	86 <i>a</i> Rn 209 +311(31)	
87 <i>a</i> Fr 223 117(10)	88 <i>a</i> Ra 226 -	Ac- Lr	104 <i>a</i> Rf 261 -	105 <i>a</i> Db 262 -	106 <i>a</i> Sg 266 -	107 <i>a</i> Bh 268 -	108 <i>a</i> Hs 269 -	109 <i>a</i> Mt 270 -										
			57 <i>a</i> La 138 +450(20) 139 +200(10)	58 <i>a</i> Ce 140 -	59 <i>a</i> Pr 141 -58.9(42)	60 <i>a</i> Nd 143 -630(60) 145 -330(30)	61 <i>a</i> Pm 147 740(200)	62 <i>μa</i> Sm 147 -259(26) 149 +75(8) 149 +1014(98)	63 <i>μ</i> Eu 151 903(10) 153 2412(21)	64 <i>μ</i> Gd 155 +1270(30) 157 +1350(30)	65 <i>μ</i> Tb 159 +1432(8)	66 <i>μa</i> Dy 161 +2507(20) 163 +2648(21)	67 <i>π</i> Ho 165 +3580(20)	68 <i>μ</i> Er 167 +3565(29)	69 <i>a</i> Tm 169 -1200(100)	70 <i>μ</i> Yb 173 +2800(40)	71 <i>μa</i> Lu 175 +3490(20) 176 +4970(30)	
			89 <i>a</i> Ac 227 1700(200)	90 <i>a</i> Th 229 4300(900)	91 <i>n</i> Pa 231 -1720(50) 231 690(170)	92 <i>μ</i> U 233 3663(8) 235 4936(6)	93 <i>μ</i> Np 237 +3886(6) 237 +3850(40)	94 <i>μ</i> Pu 239 -3345(13) 241 5600(200)	95 <i>a</i> Am 243 +4210 243 4100	96 <i>a</i> Cm 247 -	97 <i>a</i> Bk 247 -	98 <i>a</i> Cf 251 -	99 <i>a</i> Es 253 6700(800)	100 <i>a</i> Fm 257 -	101 <i>a</i> Md 259 -	102 <i>a</i> No 259 -	103 <i>a</i> Lr 261 -	

** Methods for the primary Q value: 'a' atomic, 'm' molecular, 's' solid-state, 'μ' muonic, 'n' nuclear state lifetime, 'π' pionic. 'X+Y': combines 'X' and 'Y'.
 A star (*) indicates an excited nuclear state. Underlined isotopes refer to Mössbauer states.

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Figure 2: 'Year-2001' Q values. P. Pyykkö, *Mol. Phys.* **99**, 1617-1629 (2001).