

On the nature and extent of chalcogen $\cdots\pi$ interactions

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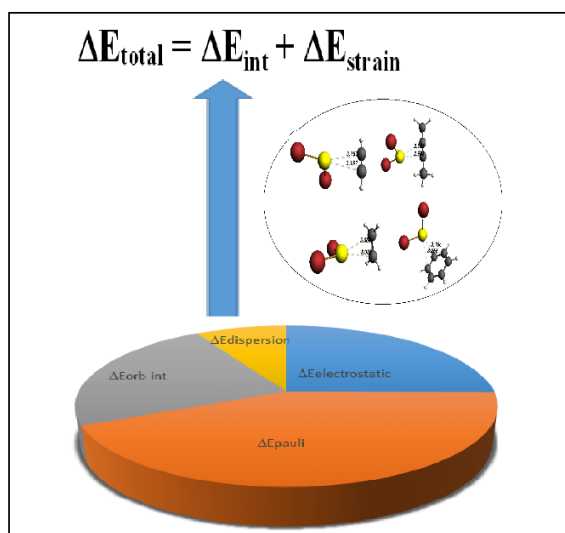
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Weak non-covalent interactions have gained paramount importance in chemistry. They play a crucial role in different phenomena, among which the self-assembly of large molecules, biological pattern recognition and crystal packing. Chalcogen $\cdots\pi$ interactions, which are the focus of the present study, were found in biological active compounds like cysteine, selenocysteine, selenomethionine.¹⁻²

In our analysis, calculations were performed to predict, characterize and quantify chalcogen $\cdots\pi$ interactions between X_2Y ($X = F, Cl, Br, I$ and $Y = S, Se, Te$) and an unsaturated substrate (ethyne, 2-butyne, ethylene and benzene). The density functional BLYP in combination with TZ2P basis set was used and scalar relativistic effects were accounted for through zeroth-order regular approximation (ZORA).³ MP2/aug-cc-PVTZ (PP) results were included for selected cases. A fragment-based approach (Activation Strain Analysis, ASA)⁴ was employed to gain insight on the nature of the interaction established between the chalcogen center and the π system, which behave like a Lewis acid-base couple. Trends are shown and discussed and in perspective will be extended to biological sites to investigate chalcogen based redox active enzymes and supramolecular aggregated architectures.

Keywords: Activation Strain Analysis, (ASA), chalcogen $\cdots\pi$ interactions, DFT calculations, ZORA, chalcogen centers



References

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